

MAGAK, Jiri; WIMMER, Petr; ZACHER, Jan

Analytic evaluation of the acid component of low-temperature tar from Most lignite, Sbor pal vod VSChT Vol. 5:251-279 '61 [publ, '62].

1. Katedra koksarenstvi a plynarenstvi, Vysoka skola chemicko-technologicka, Praha.

MACAK, Jiri

Thermal-catalytic cracking of liquid hydrocarbons. Sbor
pal vod VSChT Vol. 5:281-293 '61 [publ. '62].

1. Katedra koksarenstvi a plynarenstvi, Vysoka skola chemicko-
technologicka, Praha.

MACAK, J.

Commemorating the 60th birthday of Professor Rudolf Riedl. Chem
listy 57 no.9:1000-1001 S '63.

MACEK, Lubomir

Failures of mechanical equipment in panel factories and its maintenance. Poz stavby 11 nc.5:250-252 '63.

1. Krajske sdruzeni narodnich podniku ve stavebictvi, Ostrava.

MACEK, Milan; MICHL, Jiri

Contribution to cultivation of human diploid cells. Acta Univ.
Carol. [med.] (Praha) 10 no.7:519-539 '64

1. Ustav vyzkumu vyvoje ditete fakulty detskeho lekarstvi
University Karlovy v Praze (reditel prof. MUDr. J.Houstek,
DrSc.) a Ustav ser a ockovacich latek, odbor tkanovych kultur
(vedouci MUDr.J.Johanovsky, CSc.).

More plastic materials. p.7. (Technicke Noviny, Praha, Vol 2, No. 21, Nov 1954)

SO: Monthly list of East European Accessions (EEAL), LC Vol 4, No. 6, June 1955, Uncl

Development of the production and use of plastics in Czechoslovakia,
p. 15, ZA SOCIALISTICKOU VEDU A TECHNIKU (Přípravný výbor vědeckých
technických společností při československé akademii věd) Praha, Vol. 5,
No. 1, Jan. 1955

SOURCE: East European Accessions List (EEAL) Library of Congress,
Vol. 5, No. 12, December 1955

NOSEK, J.J.; MACAK, V.

New fibrigenous substances and their significance in medicine.
Cas. lek cesk. 90 no. 48:1437-1438. 23 Nov. 1951. (CML 21:3)

1. Of the Institute of Medical Chemistry of Palacky University,
Olomouc.

MACAK, V.

Chemical Abst.
Vol. 48
Apr. 10, 1954
Electrochemistry

(4)
Polarography of Terramycin, D. Krestnová-Tihonová,
O. Matl, and P. Šantav (Palacký Univ., Olomouc,
Czech.) Chem. Listy 47, 838-8 (1953).—The polarographic
behavior of Terramycin compared with that of colchicine
and cinchonine suggests a 4-electron reduction. M. Hudlický

a *compd.*, m. 79-83°, *epigallocatechin*, m. 345-50° (*tri-Ac deriv.*, m. 184-5°; *tri-Bz deriv.*, m. 223-5°), and a residue which diss. with H₂O gave at pH 2-3 *calchiceins*, an ether ext. contg. phenolic acids, m. 135-142°, a CHCl₃ ext. (I), after alkalization with NH₄ 2.56 g. another CHCl₃ ext. (II), and after neutralization 2.3 g. of an ext. obtained by the extn. with 2:1 CHCl₃-EtOH (III). Ext. I gave by chromatography: *calchicine* (IV), *compd. E*, *compd. D* (m. 230-3°), a *compd. O*, m. 234-6° (from AcOEt), $[\alpha]_D^{25}$ -114°, *compd. -J*, m. 186-8°, and *compd. J* (m. 274-8° (from AcOEt and Et₂O), $[\alpha]_D^{25}$ -150°). After acetylation of the mother liquids after the sepn. of *E*, were isolated *Ac deriv.* of *E*, m. 184-8°, a *compd. N*, m. 227-9° (from AcOEt), $[\alpha]_D^{25}$ -110°, refluxing with MeOH and MeONa gave a *compd.*, m. 240-5°, *Me deriv.*, (with CH₃N₃), m. 242-5°, IV, and *compd. B*, m. 265-7° (from AcOEt and Et₂O), $[\alpha]_D^{25}$ -107° (acid; by the action of MeONa, m. 224-2°, *Me ester* (with CH₃N₃), m. 174-6°). II yielded by chromatography *compd. F*, m. 134-6°, *compd. S*, m. 136-8° (from MeOH and Et₂O), $[\alpha]_D^{25}$ -120°, and, after acetylation, a *compd.*, m. 224-6° (from AcOEt and Et₂O), $[\alpha]_D^{25}$ -91° (acetylated *compd. C* or *D*). From 2.8 g. III sep'd. 65 mg. *compd. M*, m. 310-11°, *Ac deriv.*, m. 302-1°, $[\alpha]_D^{25}$ -214°. M. Hudlický

MACAK, VL.

(6)

✓ Cadmium intoxication. VL. Mařík, M. Černoch, J. Bartek, D. Wiedermann, and F. Santavý. *Lékařské Listy* 9, 27-30(1954).—The authors believe that in intoxications of this type (and possibly other heavy metals) the body does not suffer any damage through the coupling of heavy metals with SH groups of glutathione. The heavy metals are probably combined with other groups in the human body by means of more labile coupling and therefore it is possible to carry out the detoxication in a rather simple way, i.e. by means of cysteine or British antilewisite.

O. E. Lobstein

KRESTYHOVA-TELUPILOVA, O.; MACAK, V.; SANTAVY, F.

Polarography of terramycin [with summary in German]. Sbor.Chekh.khim.
rab. 19 no.2:234-237 Ap '54. (MLRA 7:6)

1. Khimicheskiy institut meditsinskogo fakul'teta universiteta im.
Palatskogo Olomouts. (Terramycin) (Polarograph and polarography)

MACAK, VLADIMIR

ref

CZECH

Substances of Colchicum autumnale and their derivatives.
XXXIV. Isolation of further substances from the flowers of
Colchicum autumnale. Frantisek Santavy and Vladimir
Macak (Palacky Univ., Olomouc, Czech). *Collection
Czechoslov. Chem. Commun.* 19, 805-15 (1954) (in German).
—See C.A. 49, 1001a. E. J. C.

MACAK, Vladimir (Olomouc, Jakoubka ze Stribra 66.)

Spontaneous pneumoperitoneum in children. Cesk. pediat. 14 no.4:349-351
5 Apr 59.

1. Detska klinika FU v Olomouci, prednosta doc. MUDr. Ant. Mores.
(PNEUMOPERITONEUM, in inf. & child.
(Cz))

MACAK, V.

Congenital stridor from anomaly of the great vessels. Cesk.pediat.
14 no.9:808-811 S '59.

1. Detske klinika PU v Olomouci, prednosta doc. MUDr. Ant.Mores.
(AORTA abnorm.)
(RESPIRATION)

RAPANT, Vladislav; MACAK, V.; DOUBRAVSKY, J.

Double aortic arch. Cas. lek. cesk. 98 no.11:326-330 13 Mar 59.

1. Chirurgická klinika PU v Olomouci, přednosta prof. dr. Vlad.
Rapant a dětská klinika PU v Olomouci, přednosta doc. dr. A. Mores
VI. R., Olomouc, chirurg. klinika.

(AORTA, abnorm.

double aortic arch, surg. & case report (Cz))

MACAK, Vladimir

Hemangioma of the entire lower extremity with thrombopenic purpura.
Cesk.pediat. 15 no.1:50-53 Ja '60.

1. Detska klinika PU v Olomouci, prednosta doc. MUDr. Ant. Mores
(HEMANGIOMA in inf.&child.)
(LEG neopl.)
(PURPURA THROMBOPENIC in inf.& child.)
(INFANT NEWBORN dis.)

MACAL, V.

The M 029 testing device of coil-winding insulation using surge voltage.

P. 17. (ELEKTROTECHNIK) (Praha, Czechoslovakia) Vol. 13, no. 1, Jan. 1958

SO: Monthly Index of East European Accession (EEAI) IC. Vol. 7, No. 5, May 1958

ARSOV, D., prof., dr.; DAVCEV, doc. dr.; MACALI, M., dr.

Survey of cases of hyperthyroidism observed from 1958-1960 in an internal clinic in Skopje. Med. glasn. 15 no.7/8:305-307 J1-Ag '61.

1. Interna klinika Medicinskog fakulteta u Skopju.

(HYPERTHYROIDISM statist)

ARSOV, D., prof. dr.; DAVCEV, P., prof. dr.; MACALI, M., dr.

Primary carcinoma of the liver in patients of the Skoplje
Internal Clinic during the past 5 years. Med.Glas.17
no.11/12:441-443 N-D '63.

1. Interna klinika Medicinskog fakulteta u Skopju.

DEJANOV, I.; TOFOVIC, P.; SERAFIMOV, K.; LAZAROV, A.; MACALI, M.

Venous thrombosis and malignant neoplasms. Acta chir. Jugosl.
12 no.1:33-41 '65.

1. Zavod za transfuziju krvi SR Makedonije (v.d. direktora dr.
N. Stojcenska), Hirurska klinika (direktor prof. dr. B. Dragojevic),
Ginekolosko-akuserska klinika (direktor prof. dr. A. Cakmakov) i
Injerna klinika (direktor prof. dr. D. Arsov) Medicinskog fakulteta
u Skopju.

NEDELKOVSKI, Jonce; MACALI, Marija; URUMOVA, Epsa

Apropos of 4 cases of chronic idiopathic jaundice (Dubin-Johnson syndrome). God. zborn. med. fak. Skopje 11:201-208 '64.

1. Interna klinika na medicinskiot fakultet, Skopje (upravnik: prof. d-r. Dimitar Arsov) i Institut za patoloska anatomija, Skopje (upravnik: prof. d-r. Dragoslav Miletic).

PTA

MACALIK, E

4

614.8 : 621 751 53

1396

Macalik E. Blowpipe Explosion.

„Wypadek z palnikiem”. Przegląd Spawalnictwa. No 5-6, 1951.
pp. 106, 1 fig.

Details of an explosion which occurred while lighting an oxy acetylene blowpipe, resulting in the bursting of both hoses and in setting fire to the acetylene bottle. The accident was due to the failure to open the oxygen supply valve prior to lighting the acetylene at the blowpipe nozzle, i. e. while the oxygen pipe was filled with acetylene. The blowpipe was of the injector type.

MACANDER, Wanda

Investigation on the effect of bacteriophage on the saprophytic acid-fast bacilli. Gruzlica 22 no.10:691-698 Oct 54.

1. Z Oddzialu Bacteriologii, kierownik: dr. J.Jakobkiewicz
z Pracowni bakteriofagowej, Kierownik: doc. dr. I.Lipska
z Miejskiej Stacji San.-Epidem. w Warszawie, dyrektor dr.
B.Mirowski, Warszawa, Nowogrodzka 82

(BACTERIOPHAGE

eff. on saprophytic acid-fast bact.)

(BACTERIA

saprophytic, acid-fast, eff. of bacteriophage)

MACANDER, W.

Studies on acid-fast bacilli in rivers receiving purified sewage. Acta microb. polon 5 no.1-2:201-205 1956.

1. Ze Stacji Sanitarno-epidemiologicznej w Warszawie.

(BACTERIA,

acid-fast bact. in rivers receiving purified sewage (Pol))

(WATER SUPPLY,

same)

(SEWAGE,

same)

MACANDER-GOLCZ, Wanda

Effect of mycobacteriophage "P17" on acid-fast bacilli isolated from lymph node pus during increased reactions following BCG vaccination. Gruzlica 27 no.12: 1197-1202 D '59.

1. Ze Stacji Sanitarno-Epidemiologicznej w Warszawie. Kierownik
Działu Badan Bakteriologicznych: dr. J. Jakobkiewicz. Kierownik
Pracowni Gruzlicy: mgr. W. Macander-Golez.
(BCG VACCINATION)
(BACTERIOPHAGE)
(MYCOBACTERIUM TUBERCULOSIS)

POPOVIC, J.; IABAN, M.; MILIETIC, V.; BJEGOVIC, M.; MACANOVIC, J.; PRASO, R.

Results of prolonged therapy with pneumoperitoneum and antibiotics.
Tuberkuloza, Beogr. 11 no. 2: 204-210 '59.

1. Institut za tuberkulozu NR Srbije, Beograd, direktor: prof. dr
M. Grujic.

(PNEUMOPERITONEUM ARTIFICIAL ther.)
(ANTITUBERCULAR DRUGS ther.)

POPOVIC, J.; MILETIC, V.; BJEGOVIC, M.; MACANOVIC, M.; ILIC, Lj.; PRASO, R.

Our results of the treatment of hematogenous pulmonary tuberculosis with combined hormones and bacteriostatic agents. Tuberkuloza, Beogr. 11 no.2:215-219 '59.

1. Institut za tuberkulozu NR Srbije, Beograd, direktor: prof. dr. Grujic.

(TUBERCULOSIS MILIARY ther.)

(TUBERCULOSIS PULMONARY ther.)

(CORTISONE ther.)

(CORTICOTROPIN ther.)

TURCU, I.; BILCU, Cl.; SCAFES, S.; MACARESCU, V.

The investigation of serum catalase in epidemic hepatitis. Stud.
cercet. inframicrobiol. 13 no.3:385-390 '62.
(HEPATITIS, INFECTIOUS) (CATALASE) (ENZYME TESTS)

MAKARESKU, V.S. [Macarescu, V.S.]; SLYUSAR', B.S.

Structural characteristics of the Neogene cover of the Epi-
Hercynian platform in Bessarabia. Izv. AN Mold. SSR. no.4:
44-60 '62. (MIRA 18:3)

MACARIE, C.

Construction of a dam for supplying water to a thermoelectric-power station. p. 583

INDUSTRIA CONSTRUCTILOR SI A MATERIALELOR DE CONSTRUCTIL, Bucuresti, Vol 6, No. 11,
Nov., 1955

SO: East European Accessions List (EEAL) Library of Congress, Vol 5, No. 7, July, 1956

MACARIE, D.

TECHNOLOGY

Periodical: REVISTA INDUSTRIEI ALIMENTARE. PRODUSE VEGETALE. No. 6, 1958.

MACARIE, D. One year of fruitful education in the food industry. p. 28.

Monthly List of East European Accession (EEAI) LC, Vol. 8, no. 3
March 1959 Unclass.

Macarie, I.

MACARIE, I.
BORN (in copy); Given Name

3

Country: Rumania

Academic Degrees: -Conf. Dr.-

School
Affiliation: Rumanian School of Veterinary Medicine (Facultatea de Medicina
Veterinara), Bucharest.

Source: Bucharest, Probleme Zootehnice si Veterinara, No 6, Aug 1961,
pp 51-62.

Date: "Parenteral treatment of Sheep Dictyocaulosis."

Co-authors:

✓ LUNGU, T., Veterinarian, School of Veterinary Medicine, Bucharest.
✓ MACARIE, I., Veterinarian, School of Veterinary Medicine,
Bucharest.

RUMANIA

Veterinarian E. POLL, Dr I. MURGU and Veterinarian I. MACAZIE, Faculty of Veterinary Medicine (Facultatea de medicina veterinara) Bucharest.

"Endocrine Pancreatic Function in Ducks."

Bucharest, Revista de Zootehnie si Medicina Veterinara, Vol 13, No 4, Apr 63; pp 95-102.

Abstract [English summary modified]: Whereas normal ducks are extremely resistant to both endogenous insulin and glucose in comparison to mammals, after pancreatectomy the insulin dose required to provoke hypoglycemic shock is paradoxically decreased (from 100 to 10 I.U.!) A hyperglycemic condition is brought about by pancreatectomy, but this is not incompatible with survival; such ducks can still handle glucose and hepatic and muscle glycogen is lower than before operation. Four graphs, 3 photographs; 3 Rumanian, 2 Western references.

1/1

RAZVAN, E., ing.; MACARIE, I., ing.

Results obtained by some studies on the dissipation of
energy. Studiul hidraul 5:301-324 '63.

BARUZI, L.; BOLOGA, V.; JURCA, I.; MACARIE, I.

Persian type glass treated with ammonium sulfate. Industria
uscara 11 no.6:322 Je '64.

1. Turda Glass Manufacture.

ROMANIA

MACANIE, Maria, Professor, Bucharest [affiliation not given]

"On the Activities of the 'Young Bee-Breeders' Circle."

Bucharest, Natura, Seria Biologie, Vol 15, No 1, Jan-Feb 1963, pp 70-74.

Abstract: Describes the activities of the "Young Bee-Breeders' Circles" organized under the auspices of the Station of Young Naturalists of the Pioneers' Palace of Bucharest in September 1961. The Circles include Pioneers and pupils in grades 3 to 7 of the schools of general education of Bucharest.
Includes 3 figures.

1/1

DIMITRIU, G.C., prof.; BULIGESCU, L., dr., candidat in stinte; SOLOMON, Sela; MACARIE, Olga; GASMET, E.

The hepatic cytolysis syndrome. Med. intern. 14 no.1:1-17 Ja '62.

1. Lucrare efectuata in Clinica medicala a Spitalului "C. Davilla".
(LIVER DISEASES)

ZAMFIR, C., dr.; medic emerit; GIOBA, Gh., dr.; EFANOV, A., dr.; ANDRONIC, C., dr.;
MACARIE, O., farm.; COMITESCU, O., farm.

On some clinico-biological aspects of atherosclerosis in persons
under 40 years of age. Med. intern. 16 no.3:291-299 Mr'64

1. Lucrare efectuata in Sectia I de boli interne si in Labora-
torul de chimie biologica din Spitalul militar central,
Bucuresti.

RUMANIA

NESTORESCU, N., Prof. Dr., Member Correspondent of the Academy of Socialist Republic of Rumania (membru corespondent al Academiei Republicii Socialiste Romania); STRATI, I., Lieutenant-Colonel, Medical Corps; and MACARIE, O., Lieutenant-Colonel, Pharmacist.

"Pleural Reaction and Local Diffusion of Penicillin in Irradiated Mice"
Bucharest, Revista Sanitara Militara, Vol 16, Special No., 1965; pp 422-425

Abstract: Effect of 300 r radiation on turpentine "pleurisy" in rats. Penicillin administration was followed by higher concentration in the pleural fluid in irradiated than in control rats. 2 tables.

RUMANIA

GIURGIU, T., Lieutenant-Colonel, Medical Veterinary Corps; IONESCU, P., Major, Medical Veterinary Corps; SAFTA, T., Major, Medical Veterinary Corps; MACARIE, O., Lieutenant-Colonel, Pharmacist; and NICULESCU, Aurelia, Chemist

"Experimental Study on Humoral and Functional Changes in the Syndrome Caused from Combined Action of Burns and Whole Body Irradiation"

Bucharest, Revista Sanitara Militara, Vol 16, Special No.. 1965; pp 466-473

Abstract: Study on 24 dogs, X-irradiated 500 r; burns 1 or 2 hrs., 2 1/4 hrs. or 5 days later (second and third-degree, 15 or 16% of the body surfaces). The 10 irradiated and burned dogs died after 8 to 11 days; of the dogs which were only burned, 55 % died after 15 to 22 days; those irradiated only all died after 12 to 15 days. Hematologic data are diagrammed and discussed in great detail. 11 graphs, 3 kymograms.

L 30141-66
ACC NR: AP6020325

SOURCE CODE: RU/0012/65/061/001/0015/0022

AUTHOR: Zamfir, C. (Doctor; Major general); Efanov, Al. (Doctor; Major); Cioba, Gh. (Doctor; Major); Macarie, O. (Pharmacist; Major); Ionescu, M. (Pharmacist; Colonel)

ORG: none

TITLE: Concerning the atherosclerotic-type disturbances of lipid metabolism in young people with essential arterial hypertension

SOURCE: Revista sanitara militara, v. 61, no. 1, 1965, 15-22

TOPIC TAGS: biologic metabolism, blood pressure, circulatory system disease

ABSTRACT: To study the disturbances in lipid metabolism of the atherosclerotic type in essential arterial hypertension in persons under 40 years old, the authors examined the lipid balance in 97 patients with essential hypertension, of whom 62 were between 18 and 24 years and 35 between 25 and 40 years. The control group consisted of 30 patients between 40 and 51 years suffering from clinical and humoral atherosclerosis. The various clinical and laboratory tests confirmed that essential arterial hypertension induces changes in the lipid balance, even in the beginning stages of the disease and in young persons. Orig. art. has: 1 table. [JPRS]

SUB CODE: 06 / SUBM DATE: none / ORIG REF: 009 / OTH REF: 003
SOV REF: 002

Card 1/1 Trm

RUMANIA

IACOMI, D., Pharmacist, Col, MACARIE, O., Pharmacist, Lt-Col,
CONICESCU, O., Pharmacist, Lt-Col, ILIESCU, E., Pharmacist,
Lt-Col, IONESCU, M., Pharmacist, Col, and BRUJA, N., Pharmacist,
Lt-Col [affiliation not given]

"Rapid and Simplified Methods Used in Biochemistry. Preparation
of Paper Strip Tests for the Deceleration of Urinary Glucose."

Bucharest, Revista Sanitara Militara, Vol 62, No 4, Jul-Aug 66,
pp 727-734.

Abstract: The authors describe the method of preparing and
preserving paper strip kits for urinary glucose tests. The
strips are impregnated with glucoso-oxidase or peroxidase and
ortho-toluidine. The simplicity and rapidity of using such
strips and their specificity and relatively good sensitivity
are emphasized.

Includes 5 references, of which one German, 2 French and
2 English-language. -- Manuscript submitted 25 January 1966.

1/1

MARINESCU, B., Dr, Col, TEODORESCU, C., Dr, Lt-Col, MIHALACHE, Gh., Dr, Lt-Col, ANTON, M., Dr, Lt-Col, IACOMI, D., Pharmacist, Col, IONESCU, M., Pharmacist, Col, and MACARIE, O., Pharmacist, Lt-Col [affiliation not given]

Studies on the Lipid Spectrum of Plasma in Some Categories of Soldiers."

Bucharest, Revista Sanitara Militara, Vol 62, No 4, Jul-Aug 66, pp 683-690.

Abstract: The authors studied the plasma lipid spectrum of a group of 224 healthy soldiers, average age 32 years, divided into 3 groups according to age and diet pattern. They found that the average range of cholesterolemia in the age range 25 to 35 years is normally 150 to 200 milligram-percent, with values above 250 definitely abnormal. A direct causal relation was found between the percentage of lipid calorigenic tropines in the diet and changes in the lipid spectrum of the plasma.

Includes 4 tables, 4 figures and 17 references, of which 7 Rumanian, 4 French and 6 English-language. -- Manuscript submitted 18 November 1965.

1/1

- 10 -

MACARINI, N.; OLIVA, L.

Pneumostratigraphy of the pancreas. *Magy. radiol.* 9 no.2:69-77 July 57.

1. A genovai egyetem rontgenintezetenek kozlomenye (igazgato: prof.
A. Vallebona).

(PANCREAS, radiography
pneumostratigraphy (Hun))

24

THE CLASS OF AMMINES OF THIOCYANATES WITH BENZIDINE AND TOLIDINE. IV. G. SQUEN and C. (H. M. G. SQUEN), *Bull. soc. chim.* (7) 7, 237-47 (1933), cf. C. A. 27, 2100. Benzidine and tolidine derivs. of Co, Ni, Cd, Mn, Zn and Mg thiocyanates are described. They were synthesized by using alc. solns. of the constituents. An alc. soln. of Ni-(NO)₂·6H₂O and KCNS on treatment with benzidine (Bd) yields [NiBd₂(H₂O)₂(SCN)₂·xH₂O, where x = 1, 2, 3 or 4 according to the conditions of drying. [NiBd₂(H₂O)₂(SCN)₂ was obtained by holding the material at 75° over P₂O₅. Cd(SCN)₂ gave [CdBd₂(SCN)₂], completely anhyd. [ZnBd₂(H₂O)₂(SCN)₂·1.5H₂O and the anhyd. salt were prepd. [MnBd₂(SCN)₂·H₂O was obtained. In all the above compds. benzidine occupies 2 coordination positions. In the Mg compds. [MgBd₂(H₂O)₂(SCN)₂·4H₂O, [MgBd₂(H₂O)₂(SCN)₂ and [MgBd₂(SCN)₂] the base occupies a single coordination position. The compd. contg. two mols. of H₂O can be left for days in contact with pyridine, with no change, but the anhyd. form adds two mols. of pyridine. A similar series of tolidine compds. was prepd. with the exception of the Mg compd. In these the base occupies a single coordination position, and the H₂O is more loosely bound than in the corresponding benzidine compds. None of the compds. is sol. in cold H₂O, and all are decomposed by hot H₂O. Most of them are sol. in EtOH and Me₂CO, and all are sol. in Et₂O, CHCl₃ and C₆H₆. The SCN compds. do not agree in all behavior with the SeCN compds. but are, on the whole, similar. C. E. P. Jeffreys.

ASM-SLA METALLURGICAL LITERATURE CLASSIFICATION

m

9

*A New Volumetric Method for the Indirect Determination of Zinc. G. Spacu and C. Gh. Macarovici (*Bol. Soc. Stiinta Cluj*). 1934, 8, 129-139; (*Chem. Zentr.*, 1935, 106, I, 1422-1423).—The method depends on the precipitation of the Zn by addition of a known volume of 0.1N-NH₄CNS solution and C₆H₅N, filtration of an aliquot part of the solution, and titration of the excess of NH₄CNS with AgNO₃ after neutralization of the solution with HNO₃. —A. R. P.

ASS-SLA METALLURGICAL LITERATURE CLASSIFICATION

BC

R-3

Constitution of double salts. XX. Diammines with benzidine and tolidine. G. Seay and C. G. Macanovich (Ind. Soc. Stuttg. Cluj, 1934, 8, 283-285; Chem. Zentr., 1934, i, 3446).—By the action of tolidine (Tid) and benzidine (Bed) on the double salts $\text{CoCl}_2 \cdot 2\text{NiCl}_2 \cdot 12\text{H}_2\text{O}$, $\text{CoCl}_2 \cdot \text{CuCl}_2 \cdot 4\text{H}_2\text{O}$, and $\text{HgCl}_2 \cdot \text{CoCl}_2 \cdot 4\text{H}_2\text{O}$, the following compounds are obtained: $[\text{CoCl}_2]_2[\text{NiBed}]_2$; $[\text{CoCl}_2]_2[\text{NiBed}][\text{CoBed}]$; $[\text{CoCl}_2][\text{CuBed}]$; $[\text{CoCl}_2][\text{CoTid}]$; $[\text{HgCl}_2][\text{CoTid}]$. By the action of NH_3 on $[\text{CoCl}_2][\text{Co}(\text{C}_6\text{H}_5\text{N})_4]$, $[\text{CoCl}_2][\text{Co}(\text{NH}_2)_4]$ is formed. J. S. A.

ASM-SLA METALLURGICAL LITERATURE CLASSIFICATION

A New Method for the Colorimetric Estimation of Cobalt. G. Spacu and C. Ch. Macarovici (*Bul. Soc. Stiinte Cluj*, 1933, 3, (2), 245-250; *Chim. et Ind.*, 1937, 37, (4), 653).—Into two beakers pour respectively equal quantities (10 to 15 c.c.) of the solution under examination and of a standard solution. To each add 0.5 cm.³ of a 1% alcoholic solution of dimethylglyoxime. Agitate the solution and then add 0.2 cm.³ of 1% alcoholic solution of benzidine or toluidine. Allow to stand for 5 minutes and then compare. The colour obtained with toluidine is deeper than that with benzidine. The sensitivity is said to be about 1:4,000,000.—W. A. C. N.

1ST AND 2ND GROUPS										3RD AND 4TH GROUPS									
PROCESSES AND PROPERTIES INDEX																			
9 <i>A Method for Simultaneous Determination of Cobalt and Nickel. G. Spacu and C. Ch. Macaroyici (Bul. Soc. Stiinta Cluj, 1958, 8, (3), 444-447; Chim. et Ind., 1958, 89, 447).—In a part of the solution Ni and Co are determined together after precipitation as $[(CoPy)_4RSCN]_2$ and $[NiPy_4RSCN]_2$. In an aliquot part of the solution, the Ni content is determined by Brunck's or Liebig's method, so that the Co content can be calculated.—D. S.</i>																			
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New complex amines belonging to the group of iron and cobalt dinitrosulphates. G. SPACU and C. G. MACAROVICI (Bul. Soc. Stiinta Cluj,

1959, 9, 107-206).—By addition of various amines to an aq. solution of $\text{Na}[(\text{NO})_2\text{FeS}_2\text{O}_6] \cdot 2\text{H}_2\text{O}$ at 0° the following compounds have been prepared: *trans*- $[\text{Co en}_2\text{Cl}_2]A$ (yellowish-green), $[\text{Co en}_2]A_2 \cdot 2\text{H}_2\text{O}$ (bronze), $[\text{Co}(\text{NH}_2)_4]A_2 \cdot 4\text{H}_2\text{O}$ (brown), $[\text{Co en}_2]A_2 \cdot 4\text{H}_2\text{O}$ (brown), and $[\text{Cr en}_2]A_2 \cdot 4\text{H}_2\text{O}$ (yellowish-brown), where $A = [(\text{NO})_2\text{FeS}_2\text{O}_6]^-$. The anion A is fairly stable in solution at 0° . By mixing aq. solutions of $\text{K}_2[\text{Co}(\text{NO})_2(\text{S}_2\text{O}_6)_2]$ and of various amines and pptg. with EtOH the following compounds were obtained: *cis*- $[\text{Co en}_2\text{Cl}_2]B \cdot 2\text{H}_2\text{O}$ (violet-brown), $[\text{Co en}_2]B \cdot \text{H}_2\text{O}$ (brown), $[\text{Co}(\text{NH}_2)_4]B \cdot 3\text{H}_2\text{O}$ (bronze), $[\text{Cr en}_2]B \cdot 3\text{H}_2\text{O}$ (bronze), $[\text{Co}(\text{Co}(\text{OH})_2\text{en}_2)]B \cdot 3\text{H}_2\text{O}$ (brown), where $B = [\text{Co}(\text{NO})_2(\text{S}_2\text{O}_6)_2]^{2-}$. Both the A and B compounds decompose slowly with evolution of NO on keeping, but are stable at low temp. O. J. W.

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Class of amines with 8-hydroxyquinoline and 5,7-dibromo-8-hydroxyquinoline. O. Spacu and C. Gh. Marcavici. *Bull. rev. sci. acad. roumaine* 22, 150 (1979).

The following complex compds. of 8-hydroxyquinoline (I) and 5,7-dibromo-8-hydroxyquinoline (II) were prepd. in which I and II were anions: [Co(NH₃)₅](OC₆H₄N)₂.4H₂O, [Co(NH₃)₅SCN](OC₆H₄N)₂.5H₂O, [Co en₂(NH₃)₂](OC₆H₄N)₂.6H₂O, [Co en₂](OC₆H₄N)₂.6H₂O, [Cr(NH₃)₅](OC₆H₄N)₂.2H₂O, [Cr(CO(NH₃)₂)₂](OC₆H₄N)₂.4H₂O, [Co(NH₃)₅](OC₆H₃Br₂N)₂, [Co en₂](OC₆H₃Br₂N)₂, [Co en₂(NH₃)₂](OC₆H₃Br₂N)₂, [Co en₂SCN]₂, [Co en₂](OC₆H₃Br₂N)₂, [Cr en₂](OC₆H₃Br₂N)₂, [Cr(CO(NH₃)₂)₂](OC₆H₃Br₂N)₂, [Cr(antipyrine)]₂(OC₆H₃Br₂N)₂, [Cr(NH₃)₅](OC₆H₃Br₂N)₂. Similar compds. of Cd, Zn and Cu could not be prepd. inasmuch as I and II became part of the cation. The general method for prepg. the above compds. consists of adding a dil. soln. of the K salt of I or II to a dil. soln. of the corresponding chloride or nitrate. J. C. Lo Cicero

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The autoxidation of sunflower oil and the Tüfel-Sadler reaction. C. Gh. Macaravici. <i>Bull. sect. sci. acad. roumaine</i> 22, 495-513 (1940) (in German).—When sunflower oil was stored (in the presence of air and light) in flasks (half-filled), autoxidation of the oil took place with resulting increase in peroxide no.; the refractive index, sapon. no. and acidity of the oil increased slightly, the I no. decreased slightly, the sp. gr. remained unchanged and the Schiff reaction and the Tüfel-Sadler reaction (C. A. 28, 3607¹) were pos. The greatest change in peroxide no. took place when the oil was stored in a blue-glass flask, less in a brown- or green-glass flask and least in a clear-glass flask. Addn. of 2-5% of oat flakes (contg. no husks) to the oil increased the autoxidation during storage; CCl₄ and septasin (<i>p</i>-benzylaminobenzenesulfonamide) were also pos. catalysts for the autoxidation of sunflower oil. Destruction of the natural inhibitor by heating the oil almost to boiling for approx. 1 hr. (before																										storage) resulted in increased autoxidation of the oil, both with and without oat flakes or septasin. Rancid sunflower-oil samples, upon heating, darkened, remained rancid and showed no improvement in odor and taste, but the peroxide no. decreased to a very small value; rancid samples showed little or no improvement upon treatment with roasted kieselguhr or animal charcoal. Com. samples of sunflower oil obtained in the Cluj district were either fresh or well preserved; some of the samples had a pos. Tüfel-Sadler reaction (<i>loc. cit.</i>). From a comparison of the peroxide nos. and Tüfel-Sadler reactions of a no. of samples of sunflower oil, it was concluded that oils (which are not profoundly altered) showing a neg. Tüfel-Sadler reaction are fresh or have been stored out of contact with the air. A pos. Tüfel-Sadler reaction probably indicates that the induction period of the oil is ended.																									
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Comparison of the Tafel-Sadler test with the peroxide number (in studying the autoxidation) of fats and oils. C. Gb. Macginness. *Bull. Inst. Sci. Acad. Roumaine* 23, 372-382(1941); *Chem. Zentr.* 1941, II, 3752-3; cf. C. A. 35, 2740¹.—The autoxidation of various fats and oils was followed by the simultaneous detn. of a simplified form of the Colin Henry Lea peroxide number (I) (cf. C. A. 25, 4121) and the Tafel-Sadler-Russow unsaturation (II) (cf. C. A. 26, 8631) of the Kera test. The fats investigated were hog fat and lard, beef and young-buffalo tallow, butter, chicken fat, and the seed oils of pumpkin, hemp, rape, poppy, flax, yellow and black mustard and soybean. The animal fats, obtained directly from the slaughter house, were melted for the measurements. The vegetable oils were obtained in the lab. by extr. of the pulverized whole seeds with petroleum ether. The pumpkin-seed oil was a commercial product. Data in the form of tables and curves are discussed. The change in I with time is different for each oil. Values of I and II are characteristic of the individual oil and agree only when I is measured at the end of the induction period. Autoxidation curves provide a more reliable criterion of the stability of an oil than does I.

M. I. McGinness, Jr

ASAC 564 METALLURGICAL LITERATURE CLASSIFICATION

REGION 1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31 32 33 34 35 36 37 38 39 40 41 42 43 44 45 46 47 48 49 50 51 52 53 54 55 56 57 58 59 60 61 62 63 64 65 66 67 68 69 70 71 72 73 74 75 76 77 78 79 80 81 82 83 84 85 86 87 88 89 90 91 92 93 94 95 96 97 98 99 100

CA

72

Chromatographic detection of the adulteration of commercial paprika powder. C. Gh. Macarovic and Maria C. Macarovici. *Bull. sect. sci. acad. roumaine* 24, 313-20 (1942); *Chem. Zentr.* 1942, II, 1677. Place a layer consisting of 1.5 cc. of CaCO_3 , on top of 1 cc. of Ca(OH)_2 on cotton in a glass tube 8-10 cm. long and 8 mm. in diam. and drawn to a capillary at one end, draw in air for approx. 1 min. through the absorption column and pour simultaneously 3-4 cc. of the benzene ext. of the sample and wash 2-3 times with approx. 1 cc. of benzene; this drives artificial coloring matter into the Ca(OH)_2 layer, where it changes its color. The paprika powder is considered pure if no dark-blue-violet ring is formed in the Ca(OH)_2 layer.

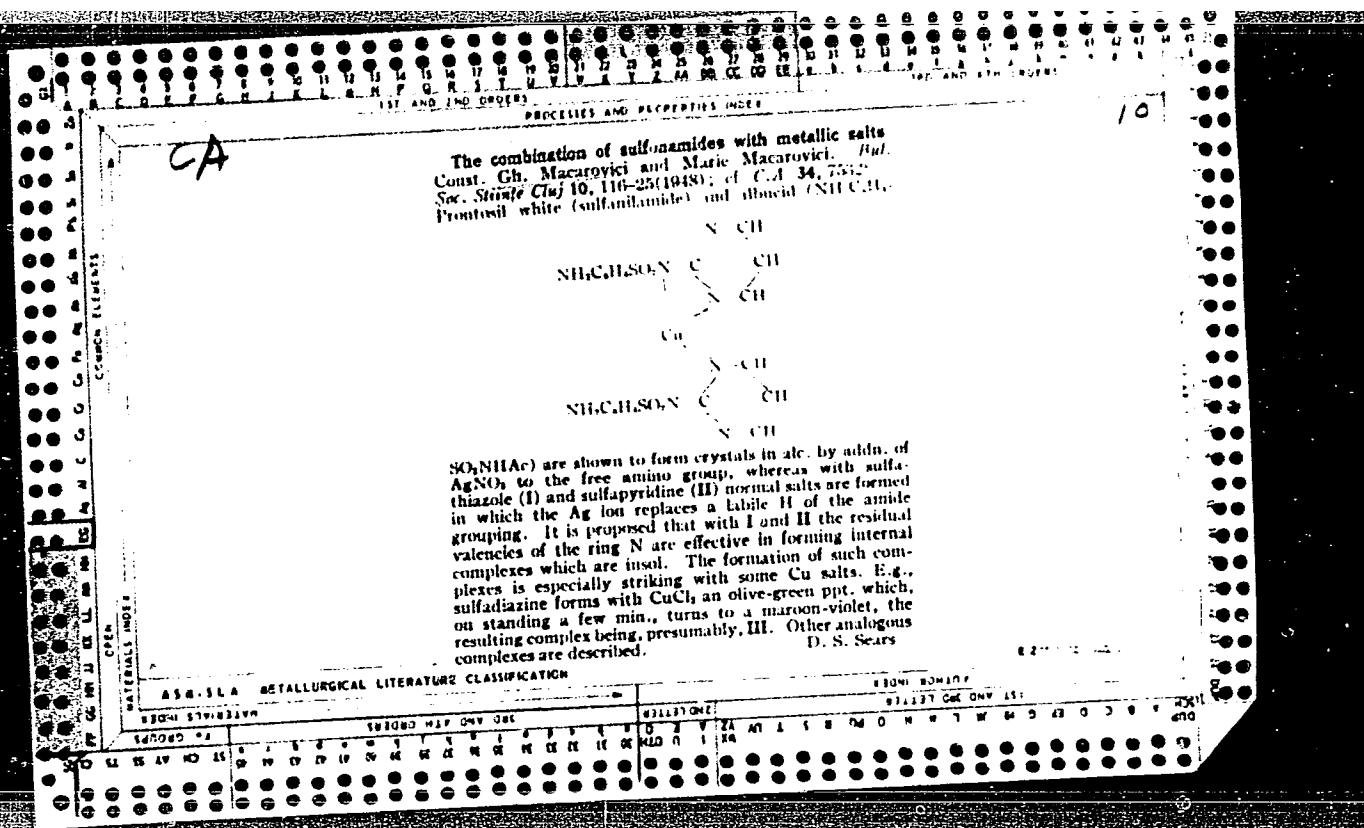
W. R. Henn

1ST AND 2ND CIPHERS		PROCESSING AND IDENTIFICATION INDEX		3RD AND 4TH CIPHERS	
CA THE determination of tellurium ion as tellurium and its separation from bismuth R. Rhen and C. G. Moore, <i>Anal. Chem.</i> (Univ. Cal.-Temerav). <i>Bull. Soc. Chim. Ind. France</i> 26, 281-84 (1944) (in French); <i>Chem. Zentr.</i> 1944, II, 945. To the slightly acid or alk. Te^{++} soln. (not more than 50 mg. Te) add 10-12 cc. concd. HCl and dil. to 150 cc. Add 3-4 g. KI, boil, add 25 cc. of satd. SO_2 soln. and boil until the colloidal Te separates. Add another 15-25 cc. of the SO_2 soln. and again boil. After a few min. decant through a filtering crucible and wash the ppt. with 2% KI soln., then 3 times with hot water, 1-2 times with cold water, and finally with alc. and ether. After drying <i>in vacuo</i> weigh the Te and subtract 4.5%. The resulting value is within $\pm 0.5\%$ of the truth. M. G. Moore					
488-514 METALLURGICAL LITERATURE CLASSIFICATION					
RECORD NUMBER 488-514					
RECORD DATE 1944					
RECORD NUMBER 488-514					
RECORD DATE 1944					

The determination of tellurium as TeO₂. C. Gh. Mac-
rovici (Univ. Cluj-Temesvar). Bull. Inst. de Chimie
16, 301-7(1940)(in French); Chem. Zentr. 1944, II, 1863.
T. preceding abstr. Tests on methods for detg. Te as
TeO₂ showed that the method of Hick and Kuta in the
presence of pyridine (cf. C.A. 20, 1588) is satisfactory.
Other org. bases with low dissov. consts. (e.g., aniline,
piperazine, or dimethylamine) can be used in place of
pyridine with equally good results. Evap. the Te soln.
to dryness, dissolve the residue in 2-6 cc. of 2 N NaOH,
and dil. to 100 cc. Add 4-5 drops of 0.1% alc. and some
bromocresol purple. Heat the soln. to boiling, and add 2
N HOAc (or K phthalate) dropwise until the color changes
to yellow (pH = 5). Boil to coagulate the ppt. After
a few hrs., filter through a filtering crucible, wash with
cold water, alc., and ether, and dry *in vacuo*. The re-
sults are about 0.35%; too low. Te can be detd. by this
method in the presence of Se. M. G. Moore

COMMON ELEMENTS
OPEN
MATERIALS INDEX
PROCESSING AND PROPERTIES INDEX
TECH. NOMENCL.

ASH-SLA METALLURGICAL LITERATURE CLASSIFICATION
TECH. SYMBOLS
RESEARCH
ANALYST'S CODE
RESEARCH
ANALYST'S CODE



The study of sulfonamides by conductometric methods
 I. The titration of sulfonamides in alcoholic solution with aqueous sodium hydroxide. Constant, Gh. Macarovici and D. Ceaparu (Univ. Cluj, Romania). *Anal. Inst. Rep. Populare Romane, Sci. Stupe. Geol., Geograf. Biol., Ser. A, 2, Mem. No. 31, 1-32 (1949) (French summary)*. The basic function of the sulfonamides being more powerful than the acidic function in aq. EtOH soln., in the equl. $RR'N \cdot C_6H_4SO_2NH_2 + H_2O \rightleftharpoons RR'N^+H \cdot C_6H_4SO_2NH_2$, the base can be displaced by conductometric titration with 0.1 N NaOH (1st inflection point). The neutralization and the formation of the Na salt takes place as soon as the sulfonamide is in its I form. When neutralization is complete, the excess of NaOH is marked by the 2nd inflection point. The hydrolysis (1st part of the curve) decreases with increasing concn. Sulfapyridine, sulfathiazole, and ultra-septyl (which have a higher acidity than sulfanilamide) were titrated in a diln. 5 times as large as that of the last. Uliron and septazine behave similarly to sulfanilamide. Albucid and $p\text{-AcNH}C_6H_4SO_2NH_2$ hydrolyze most of all the sulfonamides tested. II. The titration of sulfonamides in acetic solution with aqueous sodium hydroxide. *Ibid.* 35-50. Me_2CO was used as a solvent in order to prevent the basic hydrolysis of the amine function. The curve showed only 1 inflection point with all 8 sulfonamides. III. The titration of sulfonamides with silver nitrate. *Ibid.* 51-64. Sulfanilamide may be titrated conductometrically in caustic alc. soln. with aq. $AgNO_3$. Sulfathiazole can be titrated with aq. $AgNO_3$ in neutral EtOH soln. Sulfanilamide and albucid do not yield a ppt. when titrated with aq. $AgNO_3$ in neutral EtOH, but form an aq. product described previously (C.I. 43, 39434). Other sulfonamides do not give satisfactory results (incomplete pptn., change of color, etc.) when titrated with $AgNO_3$ in either neutral or alk. EtOH soln. Gerhard Aufleger

Sulfamides IV. Titration of sulfamides with silver nitrate in the presence of adsorption indicators. Const. Gh. Macariovici and D. Ceangulescu. *Analele Acad. Rep. Populare Romane, Ser. Mat., Fiz., Chim.* 3, Mem. 21, 1 (1950). (French summary); cf. *ibid.*, *Sect. Stunje Geol. Geograph. Biol. Ser. A*, 2, Mem. 33, 51-64 (1949). —Sulfanilamide in alk. EtOH soln. can be titrated with 0.1 N AgNO₃ with diphenylcarbazone as adsorption indicator (color of ppt. changes to blue) providing too much NaOH is not present (0.05 N alcoholic sulfanilamide, N aq. NaOH, 0.1 N AgNO₃, 8 drops 0.3% alcoholic diphenylcarbazone). Sulfathiazole (0.02 N) in neutral EtOH can also be titrated with 0.1 N AgNO₃ with a 0.5% aq. soln. of sol. eosin (C₂₀H₁₂O₆Br₂N₄) as adsorption indicator. The end point is reached when the ppt. and soln. are pink. A correction has to be applied to the result. The change in color appears 10.3% before the true end point. Error limit of the method: ± 4%. **V. Control of sulfamides used as drugs.** *Ibid.* 11, 27. It was possible to titrate not only pure sulfamides by methods described in (I), (II), (III) and (IV), but also com. products used as drugs. Dissolve the powd. compd. in EtOH or acetone and reflux for 1.5-3 hrs. Carry out the conductometric titration under the conditions described above (I), (II), and (III) do not require filtering; for titration with adsorption indicators (IV) a clear soln. is necessary. Tests were made with sulfanilamide, sulfathiazole, and sulfapyridine drugs. Gerhard Aufleger

MACAROVICI, CONST. GH.

2/
Potentiometric titration of the sulfonamides with iodic acid.
Const. Gh. Macarovici and Ion Munteanu. *Acad. rep. populare Romane, Filiala Cluj, Studii cercetari filin.* 3, No. 1/2, 24-31(1952).—Sulfonamide, sulfathiazole, ultra-septile, and sulfapyridine, react in alc. soln. with HIO_3 . Ulicon and the sulfadiazine do not give this reaction. The resulting compd. is an HIO_3 salt of the amine. HIO_3 gives with sulfanilamide the iodate. The above reaction can be used for titration for sulfanilamide in tablets.
Krikor L. Relzian

MACAROVICI, CONST. GH.

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Chen
Critical study of the volumetric analytical methods for magnesium and calcium and for magnesium in dolomite and dolomitic limestones. Const. Gh. Macarovici, Em. Leubny, A. Dorutiu, and M. Bascanu. *Rev. Chim. Cluj* (Cluj-Napoca, Romania). *Acad. rep. populare Rom. Filiala Cluj. Studii cercetari stiint.* 3, No. 3/4, 54-60 (1952). -- By studying many pertinent methods published in the literature, the following conclusions were reached: The Mdivant method (C.A. 33, 7234) gives good results, but is too time-consuming (30 hrs.). The Pyl and Samter method (cf. C.A. 19, 884) is also satisfactory, but the Babkin and Chulka method (C.A. 31, 2061) gives somewhat low values. These 3 methods can be combined to give a new method which gives good results and is not too long. Older methods quite suitable for analyses of series of samples are given by Precht (*Z. anal. Chem.* 18, 439 (1879)) and Kolthoff (cf. C.A. 12, 195), and these methods can still be improved by reverse titration, i.e., from an acid into an alk. soln., with phenolphthalein as indicator. Pptg. Mg^{++} or Ca^{++} with K_2CO_3 and titrating the excess with HCl furnishes too high results owing to adsorption, but in the presence of $EtOH$ (25%)

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MACAROVICI, CONST. CH. ...

this method, too, will furnish accurate values. The iodometric method for Mg⁺⁺ as proposed by Perepelov and Zhirnov (C.A. 30, 2518) is unsatisfactory and the indirect method for Mg and Ca from the same soln. according to Migeny (C.A. 27, 470) is satisfactory for Ca, but not for Mg. Two good methods for analyses of many samples, which however are quite time consuming, are the ones by Vasil'ev and Hendel (C.A. 28, 4670¹) and by Aleksandrov and Sevchenko (C.A. 44, 4300⁴). According to the Korzheniovskii method (C.A. 30, 4123³), if Mg and Ca are present in equal amts., or if more Mg is present than Ca, the Mg values will be too high and the Ca values too low. If Ca is present in much larger amts. than Mg, the Ca values will be correct, and the Mg values a little low, yet still within the limit of permissible error. To analyze sol. Mg and Ca salt solns. one may well choose the method by Freze (C.A. 31, 8423¹), which will always give good results when Mg and Ca are present in about equal amts. This method is not good for dolomite (D), as the values for Mg will be correct, but those for Ca too low. Mg in I can be detd. satisfactorily according to Dindur and Derhan (C.A. 44, 4372²), yet the best method for Mg in I is that by Tananayev (C.A. 43, 4801⁵), which needs 1 hr. only, where one detcs. CaO and CO₂ and calcs. the MgO. 42 references.

Werner Jacobson

2/2

RM

84035. Photomicrographs are given of the following sulfonamides and Schiff bases: sulfanilamide; salicylsulfanilamide; 4-methoxysalicylsulfanilamide; albucid; salicylsalbutal; 4-methoxysalicylsalbutal (from pyridine); sulfapyridine; 4-methoxysalicylsulfapyridine; sulfathiazole; salicylsulfathiazole; 4-methoxysalicylsulfathiazole; benzalsulfanilamide; p-chlorobenzalsulfanilamide; 1,2,5-bromosalicylsulfanilamide; 1,2,5-bromosalicylsalbutal; 1,2,5-bromosalicylsalbutal (from pyridine); 4-methoxysalicylsaldehyde; the Schiff base from sulfapyridine-HOCH₂CHO (from pyridine).

Gary Gerard

MACAROVICI, CONSTANTIN GH.

Manual de chimie analitica cantitativa; gravimetria.

Bucuresti, Rumania. Editura Technica, 1956, 291 p.

Monthly List of East European Accession (EEAI), LC. Vol. 8, No. 9 September, 1959.

Uncl.

Mistr:
4E2a(j)

7
Influence of ethylene, vinylene, and ethynylene groups
in *p,p'*-diaminodibenzyl, *p,p'*-diaminostilbene, and *p,p'*-
diaminotolan on their addition products with metallic salts.
C. Gh. Macarovici and M. Macarovici. *Rev. chim., Acad.*
Rep. Populare Roumaine 3, 18-23 (1968) (in French).—
Salts in 1:1 ratio of *p,p'*-diaminodibenzyl (I) in MeOH are
prepd. with AgNO₃ (yellow), ZnCl₂ (yellow) and CdCl₂
(white); salts of *p,p'*-diaminostilbene (II) are gray, light
yellow and light yellow, resp.; salts of *p,p'*-diaminotolan
are with CuSO₄ (red-brown), AgNO₃ (gray), ZnCl₂ (yellow-
brown), CdCl₂ (light yellow) and HgCl₂ (yellow-brown).
The salts are decompd. by H₂O, and are insol. in org. sol-
vents. The color of the salt becomes more intense for I → II
→ III. The structure of the I compds. is probably 2I.2
for (MX₂). Owen H. Wheeler

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MACAROVICH, C.Gh.

Distr: 452c

A procedure for the concentration and separation of titanium- and zirconium-containing minerals, starting with the detritus products from the dactles of the Vladeasa Mountains. C. Gh. Macarovich, E. Stoicovici, L. Cziglar, and H. Roth (Chem. Inst., Cluj, Romania). *Acta. rep. populare Romine, Studi cercari chim.* 6, 61-7 (1958) (French and Russian summaries).—The raw material is submitted to a 1st sepd. in water; by gravitation the majority of the sterile material is eliminated. The material thus enriched in minerals is submitted to a selective electromagnetic sepd. done by a special instrument conceived for that purpose. The instrument is composed of an electromagnet and a rotating disk whose distance from poles of the electromagnet can be varied, between which passes an endless band on which the mineral to be sepd. is placed. The width of this band is less than the diam. of the disk. When in operation the mineral is attracted to the disk, which upon rotation brings it outside the field, and drops from the disk. By selecting the distance between the disk and the electromagnet and also the intensity of the current, many sepd. can be made. In this case titaniferous magnetite is first sepd., then ilmenite, and finally titaniferous amphibole and Zr. A. Rothen

MACAROVICI, C. G.

COUNTRY	: Rumania	E-2
CATEGORY	: Analytical Chemistry--Analysis of inorganic substances	
ABS. JOUR.	: RZKhim., No. 21 1959, No.	74639
AUTHOR	: Macarovici, M. and Macarovici, C. G.	
INST.	: Rumanian Academy of Sciences	
TITLE	: Effect of the CH_2-CH_2 , $\text{CH}=\text{CH}$, and $\text{C}\equiv\text{C}$ Bonds in p, p'-diaminodibenzil, p,p'-diaminostilbene, and p,p'-diaminotolane on the Formation of Addition*	
ORIG. PUB.	: Rev Chim (RPR), 3, No 2, 209-212 (1958); Studii si Cercetari Chim Acad RPR Fil Cluj, 9, No 1-4,**	
ABSTRACT	: The authors report the development of semimicro- and micro-methods for the determination of $\text{Cu}(2+)$, $\text{Cd}(2+)$, and $\text{Hg}(2+)$, based on the utilization of p, p'-diaminodibenzil(I) for the precipitation of salts of these cations (in methanolic or ethanolic solution) as addition products (Communication II, RZKhKhim, 1959, No 20, 71348). The following minimum quantities can be determined by the above procedure: 0.75 mg Cu/ml, 0.26 mg Cd/ml, and 0.37 mg Hg/ml. 1-5 ml of a methanolic solution of	

CARD: 1/3 *Products with Metal Salts. III. Analytic Applications

COUNTRY Rumania
CATEGORY :

E-2

ABS. JOUR. : RZKhim., No. 21 1959, No.

74639

AUTHOR :
INST. :
TITLE :

ORIG. PUB. :

ABSTRACT : CuSO_4 or of an ethanolic solution of CdCl_2 or HgCl_2 ($\leq 0.01 \text{ M}$) is treated with 7-8 ml 0.01 M methanolic or ethanolic solution of I; after 30 min, the precipitates which form are separated by filtration through a filter [Gooch?] crucible, washed with methanol or ethanol, dried for 30-45 min at 110° , and weighed. The Hg precipitate may be washed with ethanol and ether and dried under vacuum for 15 min. The error in the determination of Cu and Cd is less than 1% and for

CARD: 2/3

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CARD: 3/3

COUNTRY : ROMANIA
 CATEGORY : General Chemistry - Organic Chemistry
 ABST. JOUR. : RSKHIS, No. 1, 1959, No. 1, 1.
 AUTHOR : Macarovici, I. and Bota, V.
 INST. : Romanian Academy
 TITLE : Study of Substituted p-Nitrobenzoyl Chlorides
 in Comparison with Benzoyl Chloride and Benzamide
 ORIG. PUB. : Studii si Cercetari Chim. Acad. Rom. 1959
 Cit., 1959, 9, No 1-4, 137-144
 ABSTRACT : The esterification reaction (CI) of p-nitrobenzoyl chloride (II) and p-nitrobenzamide (III) with p-nitrophenol, p-nitroaniline, and p-nitrobenzylamine, with the influence of temperature and concentration of the reagents, was studied. The results show that the rate of reaction increases with the increase of the concentration of the reagents and with the increase of the temperature. The esterification of p-nitrobenzoyl chloride (II) and p-nitrobenzamide (III) is a first-order reaction with respect to the concentration of the reagents. The rate of reaction is also influenced by the nature of the solvent. In the case of p-nitrobenzoyl chloride (II) the rate of reaction is higher in the case of p-nitrobenzamide (III) and p-nitrobenzylamine (IV). The esterification of p-nitrobenzoyl chloride (II) and p-nitrobenzamide (III) is a first-order reaction with respect to the concentration of the reagents. The rate of reaction is also influenced by the nature of the solvent. In the case of p-nitrobenzoyl chloride (II) the rate of reaction is higher in the case of p-nitrobenzamide (III) and p-nitrobenzylamine (IV). The esterification of p-nitrobenzoyl chloride (II) and p-nitrobenzamide (III) is a first-order reaction with respect to the concentration of the reagents. The rate of reaction is also influenced by the nature of the solvent. In the case of p-nitrobenzoyl chloride (II) the rate of reaction is higher in the case of p-nitrobenzamide (III) and p-nitrobenzylamine (IV).
 CARD: 1/3

COUNTRY : Rumania
CATEGORY :

G-1

ABB. JOUR. : RZKhim., No. 19, 1950, No. 67819

ABST. :
AUTH. :
TITLE :

ORIG. PUB. :

ABSTRACT : Solutions, with hydrochloric acid, there is formed $p\text{-H}_3\text{C}_6\text{H}_4\text{CONH}_2\text{NH}_2$. On CT of I in alcohol solution with aqueous HCl, there are formed two hydrochlorides (HC), one such as in the case of III, and in contradistinction to II, which yields only one HC. On CT, with aqueous NaOH, of m- and p-I, in aqueous-acetone solution, there are formed the corresponding Na-salts, and in the course thereof the NH_2 -group is not hydrolyzed. Under these conditions o-I yields $(\text{CH}_3)_2\text{C}=\text{NC}_6\text{H}_4\text{CONH}_2\text{N}=\text{C}(\text{CH}_3)_2$. The curves of CT of I in aqueous-acetone solution with aqueous HCl indicate the formation of one HC in the case of m- and p-I; this proves that the amine function is blocked. CT with aqueous HCl in

CARD: 2/3

RUMANIA/Chemical Technology. Chemical Products and Their
Application. Ceramics. Glass. Binding Materials.
Concrete.

H-13

Abs Jour: Ref Zhur-Khin., No 2, 1959, 5400.

Author : Macarovici, Const. Gh.; Basceanu, M.

Inst :

Title : Concerning the Thermal Dissociation of Rumanian Dolomite.

Orig Pub: Rev. chim., 1958, 9, No 4, 207-213.

Abstract: No abstract.

Card : 1/1

MACAROVICI, CONST. G.H.

Distr: 4E3b

Complex compounds with biphenyl derivatives. I. Complex compounds of the metallic salts of organic acids with benzidine. Const. Gh. Macarovici and G. Schmidt, Acad. rep. populare Romine, Filiala Cluj, Studii cercetari chim. 10, 59-65 (1959).—The reactions of benzidine (Bzd) and metallic salts of org. acids to form complex compds. studied. An attempt was made to stabilize and sep. from soln. the complex anions of the org. acids and to det. their usefulness in analysis. The following were prepd.: [Cu(Bzd)(H₂O)₂](AcO)₂ (I), dark-green, hexagonal crystals, m. 162-5° (decompn.); [CuBzd(EtOH)₂](AcO)₂, light-brown crystals, m. 164-6° (decompn.) (by standing in air or in vacuo this compd. loses alc. easily and changes to [CuBzd](AcO)₂, green, m. begins at 158° and at 163° it decomp.); [CuBzd](ClCH₂COO)₂, almost black, cryst. substance, m. 151-7° (decompn.); [Bzd.H₂][Cu(AcO)₂], brown needles, m. begins at 160°, at 188° it decomp.; [Bzd.H₂][Zn(AcO)₂], white needles, m. 156-8° (decompn.); [Bzd.H₂][Cd(AcO)₂], colorless needles changing to bright pink when washed with Et₂O and to yellow after standing in air, m. 151-2° (decompn.); [Bzd.H₂][Hg(AcO)₂], white crystals, m. 130-7° (decompn.); [Pb.Bzd][Pb(AcO)₂], needle-shaped, colorless crystals, m. 160°, 176-8° decompn.; [Bzd.H₂][Cu(ClCH₂COO)₂], olive-green crystals, m. 156-8° (decompn.); [Bzd.H₂][Cu(ClCH₂COO)₂], brown, needle-shaped crystals, m. 148-9°. [Bzd.H₂][Hg(ClCH₂COO)₂], gray-yellow needle-shaped crystals, decomp. on warming; [Bzd.H₂][UO₂(AcO)₂], yellow crystals, 160° (decomp. without melting); [Bzd.H₂][UO₂(ClCH₂COO)₂], fine, needle-shaped, bright-yellow crystals, 160° (decomp. without melting); the I was prepd. by Spacu (C.A. 22, 1939) but the crystal form and the color of this compound was different. All the complexes were prepd. by direct combination of benzidine, the salt of the metallic element, and the anion of the desired organic acid in an alc. medium.

A. Goldstein

MACAROVICI, Const. Gh.; SCHMIDT, G.

Study of the complex combinations with diphenylic derivatives. II.
Complex combinations of mono-, di-, trichloroacetic acids,
propionates and butyrates of Cu, Hg, Pb with benzidine. Studii
denc chimie Cluj 10 no.2:245-250 '59. (EEAI 9:9)

1. Universitatea "V.Babes" - Cluj, Catedra de chimie generala. 2.
Membru corespondent al Academiei R.P.R. (for Macarovici).

(Phenyl group) (Chloroacetic acid)
(Dichloroacetic acid) (Trichloroacetic acid)
(Complex compounds) (Propionates) (Copper)
(Mercury) (Lead) (Butyrates) (Benzidine)

MACAROVICI, Const. Gh.

Changing the reference basis for determination of atomic
weights of chemical elements. Rev chimie Min petr 13
no.8:449-451 Ag '62.

MACAROVICI, Const. Gh.; MICU-SMENIUC, Rodica

Study of complex combinations with diphenylic derivatives. Pt.8.
Studia Univ B-B S Chem 8 no.1:151-161 '63

1. "Babes-Bolyai" University, Cluj.

ROTH, Hannelore; CZEGLEDI, L.; MACAROVICI, Const. Gh.

Composition and stability of complexes of Zr(IV) and Hf
(IV) with D-gluconic and D-saccharic acids. Studii
cerc chimie Cluj 14 no.1:69-80 '63.

1. Institute of Chemistry, Rumanian Academy, Cluj Branch.
2. Corresponding Member of the Rumanian Academy (for
Macarovici).

MACAROVICI, Const, Gh.; PERTE, Eugenia; MOTIU, Emilia

Study on the complex combinations with diaminodiphenylic derivatives. Pts. 10, 12. Studii cerc chimie Cluj 14 no.2:243-263 '63.

1. Institute of Chemistry, Rumanian Academy, Cluj Branch.
2. Corresponding Member of the Rumanian Academy (for Macarovici).

MAIAROVICI, Const. G.; SCHMIDT, G.; HAMBURG, Erika.

Study of complex combinations of diphenyl derivatives. I. G.
Rev chimie Roum 9 no 1:51-63 Ja '64

1. Laboratory of General Chemistry, University of Cluj.

MACAROVICI, Const. Gh.; MICU-SEMENTUC, Rodica

Study of complex combinations of diphenyl derivatives. Pt.II.
Rev. chimie Roum 9 no.2:127-130 F '64

1. Laboratory of General Chemistry, University of Cluj.

MACAROVICI, Const. Gh.; SCHMIDT, G.

Complex combinations with diphenylic derivatives. It. 14. Rev chimie
Roum 9 no.11:693-702 N '64.

1. Chair of General Chemistry, Cluj University.

MACAROVICI, Const. Gh.; SCHMIDT, G.; HAMBURG, Erika

Study of the complex combinations with diphenylic
derivatives. Pt. 9. Studii cerc chim 12 no. 1:
4960 Ja '64.

1. Department of General Chemistry of the Babes-Bolyai
University, Cluj.

MACAROVICI, Const. Gh.; SCHMIDT, G.

Studii on the complex combinations with diphenylic derivatives.
Pt. 13. Studii cerc chim 13 no.6/7:405-411. Jg-Jl '64.

1. Chair of General Chemistry, "Eabes-Bolyai" University, 11
Arany Janos St., Cluj.

MACAROVICI, Const. Gh.; CZEGLEDI, L.

Obtained thorium combinations with D-gluconic and D-saccharic acids. Studii cerc chim 13 no.6/7:419-431 Je-Jl '64

1. Institute of Chemistry of the Rumanian Academy, Cluj Branch,
59-65 Donath St.

MACAROVICI, C.Gh.; SCHMIDT, G.

Study of complex combinations with diphenylic derivatives.
Pt.14. Studii cerc chim 13 no.11:733-740 N '64.

1. Chair of General Chemistry, "Babes-Bolyai" University,
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MACAROVICI, Const. Gh.; SCHMIDT, G.

On the complex combinations with phenyl derivatives. Pt.13.
Rev chimie Roum 9 no.6/7:397-404 Fe-Jl '64

1. Chair of General Chemistry, "Babes-Bolyai" University, 11
Arany Janos St., Cluj.

MACAROVICI, Const. Gh.; CZEGLEDI, L.

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1. Institute of Chemistry of the Rumanian Academy, Cluj Branch, 59-65 Donath St.

MACAROVICI, Const. Gh.; BOTA, V.

Study on sulfamides. Pt. 18. Studia Univ B-B S Chem 8 no.1:
163-177 '63

1. "Babes-Bolyai" University, Cluj.

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derivatives. Pt. 6. Studia Univ B-B S. Chem 8 no. 2:
75-83 '63.

MACAROVICI, Const. Gh.; SCHMIDT, Gerhard

Study of the complex combinations with diphenylic derivatives.
Pt.8. Studia Univ B-B S Chem 8 no.1:139-149 '63

1. "Babes-Bolyai" University, Cluj.

MACAROVICI, D.

15
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The thermal formation and decomposition of apatite.
I. The thermal synthesis of apatite. C. Litcanu and D.
Macarovic (Inst. Chem., Cluj, Roumania). Acad. rep.
populare Romine, Studii cercetari chim. 8, 199-200 (1957).
The formation of apatite has been studied as a function of
temp. and the length of heating. With differential thermal
analysis it was possible to establish the following succession
of reactions occurring during the heating of a stoichiometric
mixture of CaHPO_4 , CaCO_3 , and CaF_2 : $2\text{CaHPO}_4 = \text{Ca}_2\text{P}_2\text{O}_7$
+ H_2O ; $\text{CaCO}_3 = \text{CaO} + \text{CO}_2$; $\text{Ca}_2\text{P}_2\text{O}_7 + \text{CaO} = \text{Ca}_3(\text{PO}_4)_2$
+ CaO ; $3\text{Ca}_3(\text{PO}_4)_2 + \text{CaF}_2 = 2\text{Ca}_5(\text{PO}_4)_3\text{F}$. The forma-
tion of apatite has been studied via the soly. detn. of the
phosphate ion under standard conditions in 2% citric acid
soly. In order to obtain comparative data the soly. of the
phosphate from Kola apatite, tricalcium phosphate ppt.,
and a mixture of CaHPO_4 and CaCO_3 heated at 1000° and
 1200° for various intervals of time has been detd. in 2% citric
acid soln. In comparing the data obtained it is concluded
that the presence of CaF_2 has a decided influence on the in-
solubilization of the phosphate. This very strong insol-
ubilization occurs even after 2.5 min. of heating. It is as-
sumed that when the soly. of the phosphate decreases to less
than 3%, the main component is apatite. It was also
noticed that when the mixture is heated for more than 360
min. at 1200° or 240 min. at 1300° , the soly. of the phos-
phate increases. This may be caused by a disaggregation of
the apatite crystal system.
A. Bedia

GLW

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LITEANU, C.; MACAROVICI, D.

Formation and thermal decomposition of apatite. Studii
cerc chimie Cluj 14 no.1:81-124 '63.

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MACAROVIC

COUNTRY : Rumania
CATEGORY : Inorganic Chemistry. Complex Compounds.
ABS. JOUR. : RZKhim., no. 22 1959, No. 78031
AUTHOR : Macarovici, G.
INST. : Not given
TITLE : Reform or Standardization of Nomenclature in
Inorganic Chemistry?
ORIG. PUB. : Standardizarea, 11, No 1, 11-23 (1959)
ABSTRACT : No abstract.

CARD: 1/1

1ST AND 2ND COLUMNS										3RD AND 4TH COLUMNS									
MACAROVICI M. C.																			
CA																			
Chromatographic detection of the adulteration of commercial paprika powder. C. Gh. Macarovici and Maria C. Macarovici. <i>Bull. sect. sci. acad. roumaine</i> 24, 315-20 (1942); <i>Chem. Zentr.</i> 1942, II, 1977. - Place a layer consisting of 1.5 cc. of CaCO_3 on top of 1 cc. of Ca(OH)_2 on cotton in a glass tube 8-10 cm. long and 8 mm. in diam. and drawn to a capillary at one end, draw in air for approx. 1 min. through the adsorption column and pour simultaneously 3-4 cc. of the benzine ext. of the sample and wash 2-3 times with approx. 1 cc. of benzine; this drives artificial coloring matter into the Ca(OH)_2 layer, where it changes its color. The paprika powder is considered pure if no dark blue-violet ring is formed in the Ca(OH)_2 layer. W. R. Henn																			
AYB-61A METALLURGICAL LITERATURE CLASSIFICATION																			
1ST AND 2ND COLUMNS										3RD AND 4TH COLUMNS									

COMMON ELEMENTS		METALLURGY LITERATURE CLASSIFICATION	
1	2	3	4
1	2	3	4
5	6	7	8
9	10	11	12
13	14	15	16
17	18	19	20
21	22	23	24
25	26	27	28
29	30	31	32
33	34	35	36
37	38	39	40
41	42	43	44
45	46	47	48
49	50	51	52
53	54	55	56
57	58	59	60
61	62	63	64
65	66	67	68
69	70	71	72
73	74	75	76
77	78	79	80
81	82	83	84
85	86	87	88
89	90	91	92
93	94	95	96
97	98	99	100

MACAROVICI

CA

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THE COMBINATION OF SULFONAMIDES WITH METALLIC SALTS

Const. Gh. Macarovici and Marie Macarovici. *Bul. Soc. Stiinta Cluj* 10, 116-25(1948); *C.A.* 34, 7512.

Protosil white (sulfanilamide) and albumin (NH₂CH₂COOH).

$$\begin{array}{c}
 \text{N-CH} \\
 | \\
 \text{NH}_2\text{CH}_2\text{COOH} \\
 | \\
 \text{Cu} \\
 | \\
 \text{NH}_2\text{CH}_2\text{COOH}
 \end{array}$$

SO₂NHAc) are shown to form crystals in alc. by addn. of AgNO₃ to the free amino group, whereas with metallic salts (I) and sulfapyridine (II) normal salts are formed in which the Ag ion replaces a labile H of the amide grouping. It is proposed that with I and II the residual valencies of the ring N are effective in forming internal complexes which are insol. The formation of such complexes is especially striking with some Cu salts. E.g., sulfadiazine forms with CuCl₂ an olive-green ppt. which, on standing a few min., turns to a maroon-violet, the resulting complex being, presumably, III. Other analogous complexes are described.

D. S. Sears